

Rare Earths and Radioactivity

a backgrounder prepared by
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Temiscamingue May 13 2014

Canadian Coalition for Nuclear Responsibility

www.ccnr.org

Fact #1

**All Rare Earth
Mining Produces
Radioactive Wastes**

Rare Earth Elements

There are 92 naturally occurring elements, listed according to the “atomic number”, ranging from the lightest to the heaviest. Element number 1 is hydrogen. Element number 92 is uranium.

Uranium is radioactive, and it has two dozen “progeny” that are also radioactive. They are called the uranium “decay products”.

Elements numbered 57 to 71 are called “rare earths”. They are not radioactive, but they have a “chemical affinity” with uranium. (Elements 21 and 39 are also often considered to be rare earths.)

Wherever you find rare earths in nature, you find uranium as well. It is impossible to mine the rare earths without bringing uranium and its two dozen “decay products” to the surface.

The ore is radioactive even though the rare earths are not. So, when rare earths are mined, radioactive waste is left behind.

Rare Earth Elements (1)

| Z | Symbol | Name | Etymology | Selected applications |
|----|--------|--------------|--|---|
| 21 | Sc | Scandium | from Latin <i>Scandia</i> (Scandinavia). | Light aluminium-scandium alloys for aerospace components, additive in metal-halide lamps and mercury-vapor lamps,[4] radioactive tracing agent in oil refineries |
| 39 | Y | Yttrium | after the village of Ytterby, Sweden, where the first rare earth ore was discovered. | Yttrium aluminium garnet (YAG) laser, yttrium vanadate (YVO ₄) as host for europium in TV red phosphor, YBCO high-temperature superconductors, yttria-stabilized zirconia (YSZ), yttrium iron garnet (YIG) microwave filters,[4] energy-efficient light bulbs,[5] spark plugs, gas mantles, additive to steel |
| 57 | La | Lanthanum | from the Greek "lanthanein", meaning <i>to be hidden</i> . | High refractive index and alkali-resistant glass, flint, hydrogen storage, battery-electrodes, camera lenses, fluid catalytic cracking catalyst for oil refineries |
| 58 | Ce | Cerium | after the dwarf planet Ceres, named after the Roman goddess of agriculture. | Chemical oxidizing agent, polishing powder, yellow colors in glass and ceramics, catalyst for self-cleaning ovens, fluid catalytic cracking catalyst for oil refineries, ferrocerium flints for lighters |
| 59 | Pr | Praseodymium | from the Greek "prasios", meaning <i>leek-green</i> , and "didymos", meaning <i>twin</i> . | Rare-earth magnets, lasers, core material for carbon arc lighting, colorant in glasses and enamels, additive in didymium glass used in welding goggles,[4] ferrocerium firesteel (flint) products. |
| 60 | Nd | Neodymium | from the Greek "neos", meaning <i>new</i> , and "didymos", meaning <i>twin</i> . | Rare-earth magnets, lasers, violet colors in glass and ceramics, didymium glass, ceramic capacitors |
| 61 | Pm | Promethium | after the Titan Prometheus, who brought fire to mortals. | Nuclear batteries |

Rare Earth Elements (2)

| Z | Symbol | Name | Etymology | Selected applications |
|----|--------|------------|--|--|
| 62 | Sm | Samarium | after mine official, Vasili Samarsky-Bykhovets . | Rare-earth magnets, lasers, neutron capture, masers |
| 63 | Eu | Europium | after Europe . | Red and blue phosphors, lasers, mercury-vapor lamps, fluorescent lamps, NMR relaxation agent |
| 64 | Gd | Gadolinium | after Johan Gadolin (1760–1852), to honor his investigation of rare earths. | Rare-earth magnets, high refractive index glass or garnets, lasers, X-ray tubes, computer memories, neutron capture, MRI contrast agent, NMR relaxation agent, magnetostrictive alloys such as Galfenol , steel additive |
| 65 | Tb | Terbium | after the village of Ytterby , Sweden. | Green phosphors, lasers, fluorescent lamps, magnetostrictive alloys such as Terfenol-D |
| 66 | Dy | Dysprosium | from the Greek "dysprositos", meaning <i>hard to get</i> . | Rare-earth magnets, lasers, magnetostrictive alloys such as Terfenol-D |
| 67 | Ho | Holmium | after Stockholm (in Latin, "Holmia"), native city of one of its discoverers. | Lasers, wavelength calibration standards for optical spectrophotometers, magnets |
| 68 | Er | Erbium | after the village of Ytterby , Sweden. | Infrared lasers, vanadium steel, fiber-optic technology |
| 69 | Tm | Thulium | after the mythological northern land of Thule . | Portable X-ray machines, metal-halide lamps, lasers |
| 70 | Yb | Ytterbium | Ytterby , Sweden. | Infrared lasers, chemical reducing agent, decoy flares, stainless steel, stress gauges, nuclear medicine |
| 71 | Lu | Lutetium | after Lutetia , the city which later became Paris . | Positron emission tomography – PET scan detectors, high refractive index glass, lutetium tantalate hosts for phosphors |



Radioactive Uranium and Thorium are always present in any ore that is rich in rare earths.

Warnings From the EPA

“Thorium-232 and Uranium-238 are rather benign, but some of the decay products can represent **a danger to the environment due to the energetic particles and gamma rays released during radioactive decay..**”

“If best management practices are not used and/or operations are not carefully monitored, rare earth element production **may put human health and the environment at risk..**”

US EPA, Investigating Rare Earth Element Mine Development, 2011

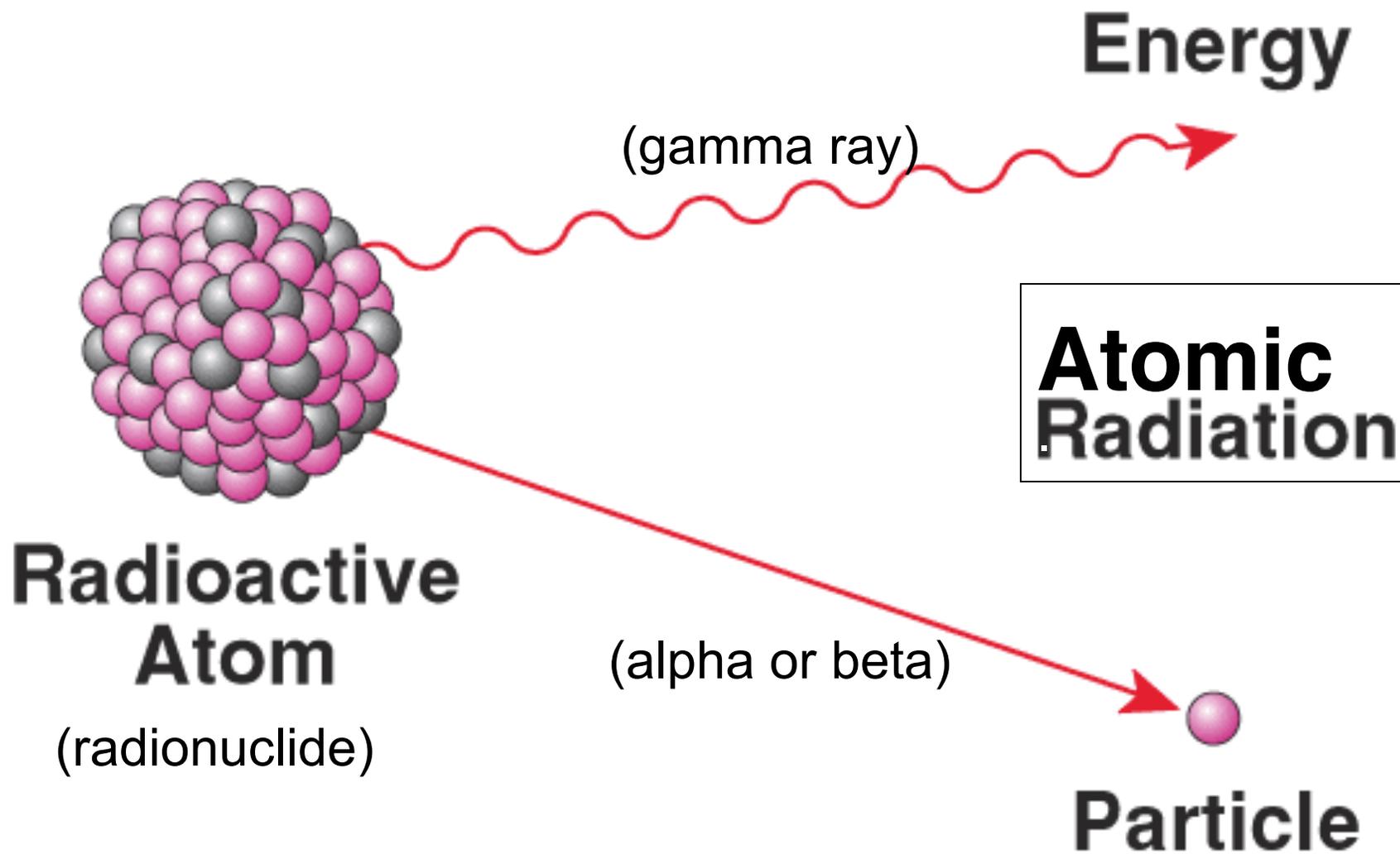


nucleus

A Model of the Uranium Atom

What is Radioactivity?

- Most materials have stable atoms (they *never* change).
- Radioactive materials have unstable atoms (they *will* change).
- Unstable atoms will **disintegrate** (suddenly and violently).
- The moment of disintegration is when biological harm is done.
- One “becquerel” indicates one disintegration (decay) per second.
- The “half-life” is how long it takes for $\frac{1}{2}$ the atoms to disintegrate.



A gamma ray is like an x-ray, but more powerful.
highly penetrating

A beta particle is like a sub-atomic bullet.
moderately penetrating

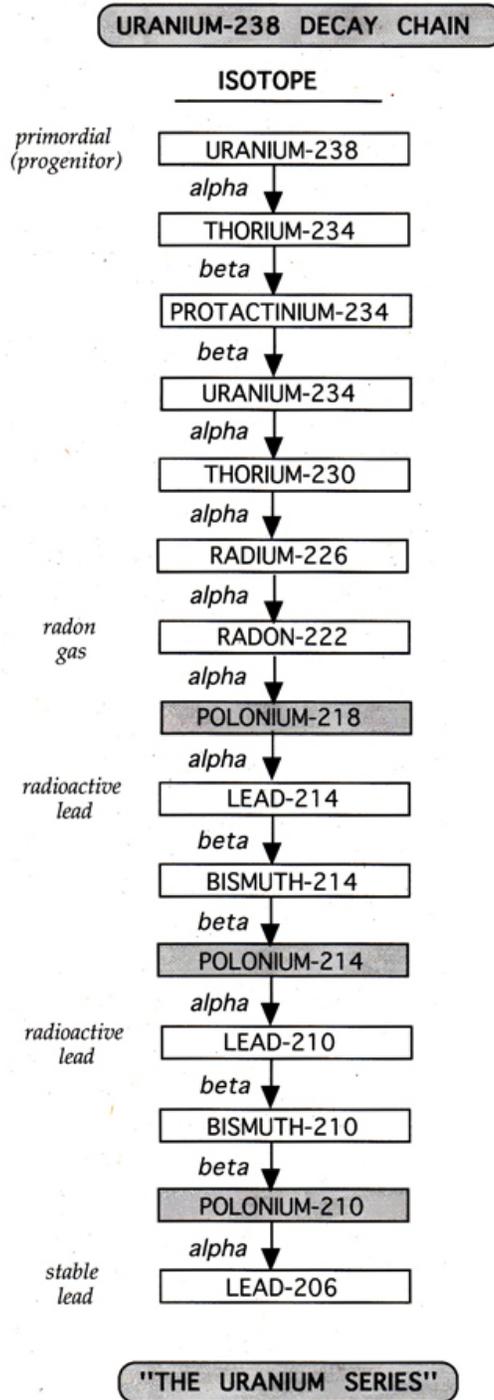
An alpha particle is like a subatomic cannon ball.
not very penetrating
~ but extremely damaging! ~

Alpha and Beta particles are INTERNAL hazards.

What is a Decay Product?

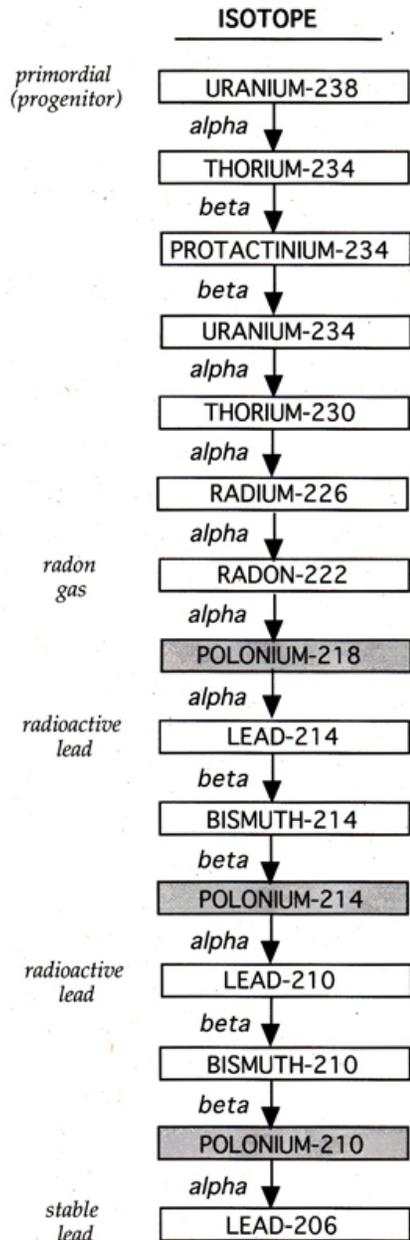
- *When disintegration (decay) occurs, a radioactive atom becomes a new kind of atom – it is **fundamentally altered**.*
- So in a radioactive material, **new atoms are being created**; they are called “decay products” or “progeny”.
- If a given decay product is also radioactive, then it will have its own decay products, and so on and so on . . .
. . . this gives rise to a “**decay chain**” or “decay series”.

Here is the decay chain of uranium-238, the most common type of uranium found in nature



It's like a family of radioactive elements – the great-great-great grandfather is uranium

URANIUM-238 DECAY CHAIN

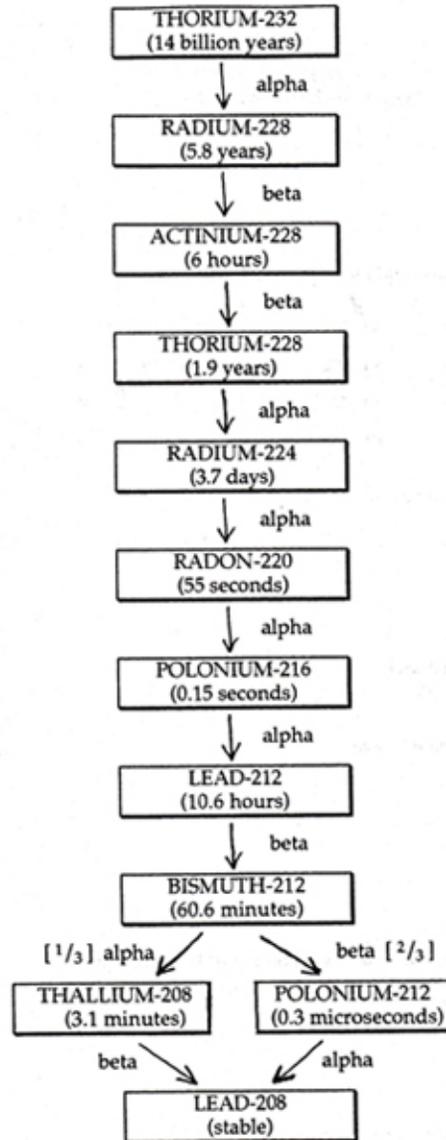


"THE URANIUM SERIES"

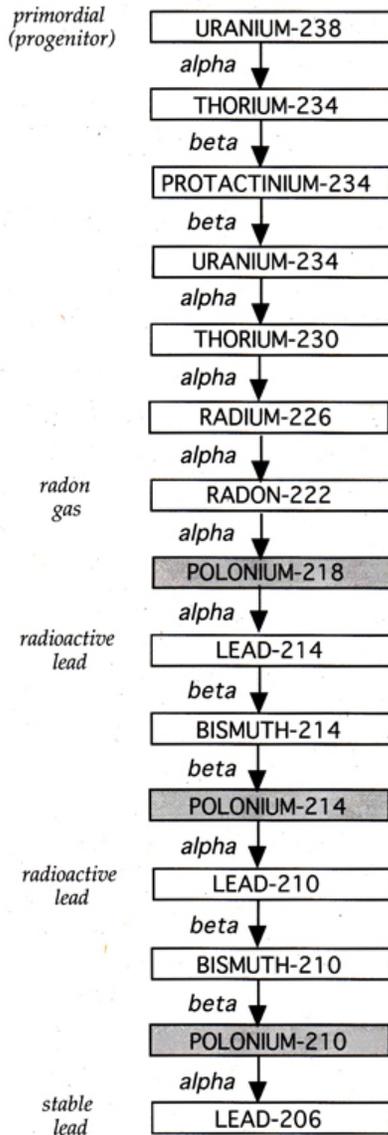
On a weight-by-weight basis, all of the uranium decay products are much more radioactive and much more dangerous than uranium itself.

The 3 types of polonium are most dangerous of all.

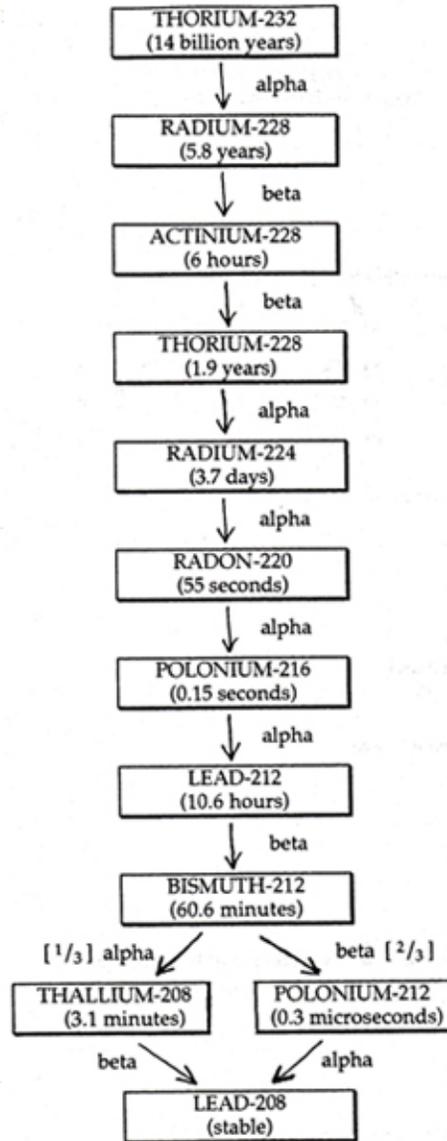
THORIUM SERIES



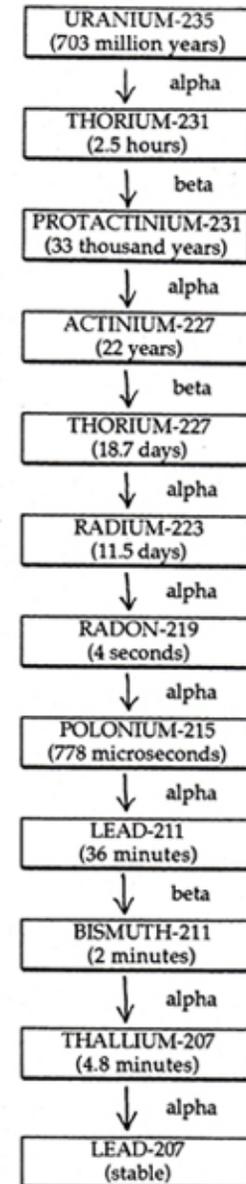
URANIUM SERIES



THORIUM SERIES



ACTINIDE SERIES



Fact #2

Radioactive
Mine Residues
Are A Health Danger

IONIZING RADIATION

THYROID

iodine-131
beta (gamma) ; 8 days

SKIN

sulphur-35
beta ; 87 days

LIVER

cobalt-60
beta (gamma) ; 5 years

OVARIES

iodine-131
beta (gamma) ; 8 days

cobalt-60
beta (gamma) ; 5 years

krypton-85
gamma ; 10 years

ruthenium-106
gamma ; 1 year

zinc-65
gamma ; 245 days

barium-140
gamma ; 13 days

potassium-42
gamma ; 12 hours

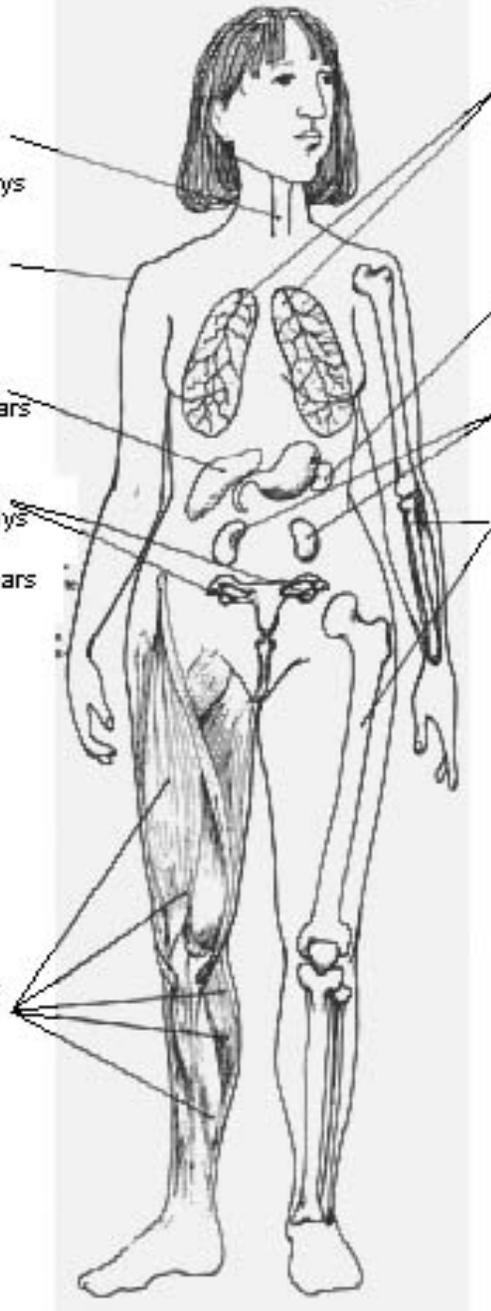
cesium-137
gamma ; 30 years

plutonium-239
alpha ; 24 000 years

MUSCLE

potassium-42
gamma ; 12 hours

cesium-137
gamma ; 30 years



LUNGS

radon-222 (and whole body)
alpha ; 3,8 days

uranium-233 (et os)
alpha ; 162 000 years

plutonium-239 (and bone)
alpha ; 24 000 years

SPLEEN

polonium-210 (and whole body)
alpha ; 138 days

KIDNEYS

uranium-238 (and bone)
alpha ; 4 500 000 years

ruthenium-106
gamma (beta) ; 1 year

BONE

radium-226
alpha ; 1 620 years

zinc-65
gamma ; 245 days

strontium-90
beta ; 28 years

yttrium-90
beta ; 64 hours

promethium-147
beta ; 2 years

barium-140
beta (gamma) ; 13 days

thorium-234
beta ; 24,1 days

phosphorus-32
beta ; 14 days

carbon-14 (and fat)
beta ; 5 600 years

Radionuclides
are chemical
substances
which are also
radioactive.

Uranium Decay Products

1. Radium - the silent killer

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Dial Painters 1920

Radium-226

Radium, like calcium, goes to bones & teeth.

Dial painters suffered from

Phase 1: Fatal anemia (blood disease)

Phase 2: Bone cancers (skeletal damage)

Phase 3: Head cancers (sinuses and mastoid)

Phase 3 : radon gas is produced in the bones by radium disintegration -- then carried by the blood to the head where the radioactive gas builds up.

In 1920s, radium sold for \$100 000 per gram.

Now it is discarded as a waste byproduct.

BCMA: "Radium is a superb carcinogen"

Uranium Decay Products

2. Polonium - the deadliest element

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Alexander Litvinenko 2006

Polonium-210

Los Alamos National Laboratory's Chemistry Division

<http://periodic.lanl.gov/elements/84.html>

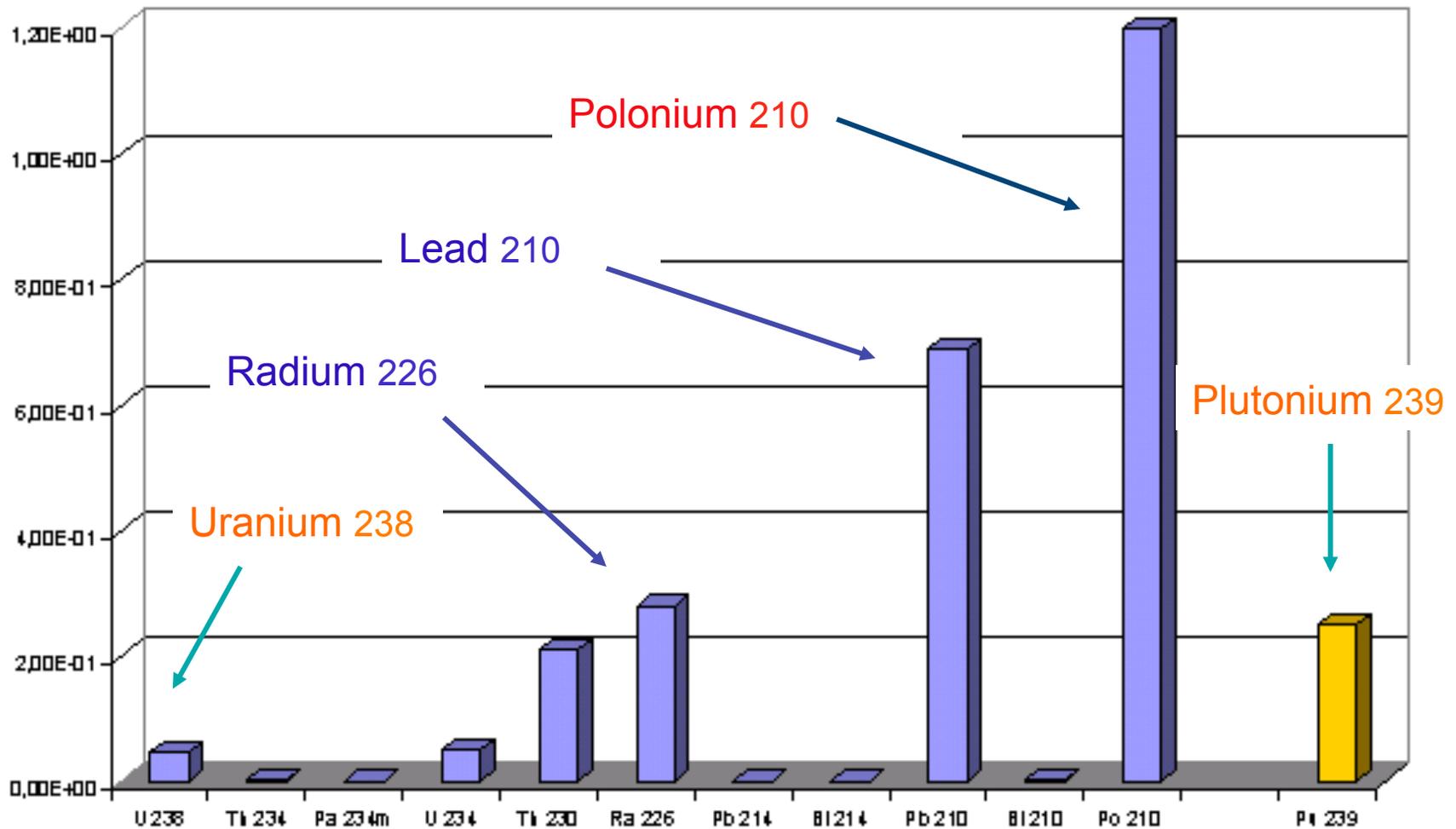
Polonium-210

Weight by weight

it is about **250 billion times**

as toxic as cyanide.

Dose Coefficients for an adult (ingestion) in microSieverts per Becquerel (official figures from EURATOM 96/29)



American Health Physics Society

Polonium-210

... is probably the cause of
up to **90 percent of the deaths**
attributed to tobacco.

(lung cancers, heart attacks, strokes)

Uranium Decay Products

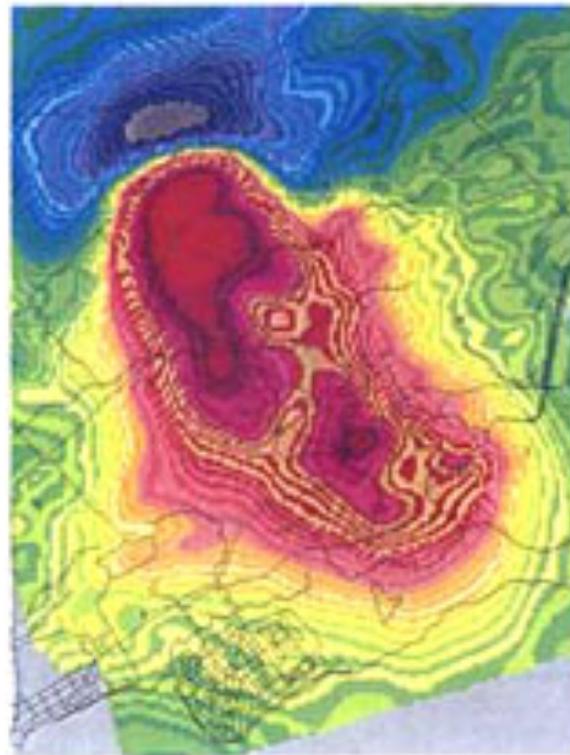
3. Radon - major public health hazard

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LE RADON À OKA

Rapport d'intervention de santé publique



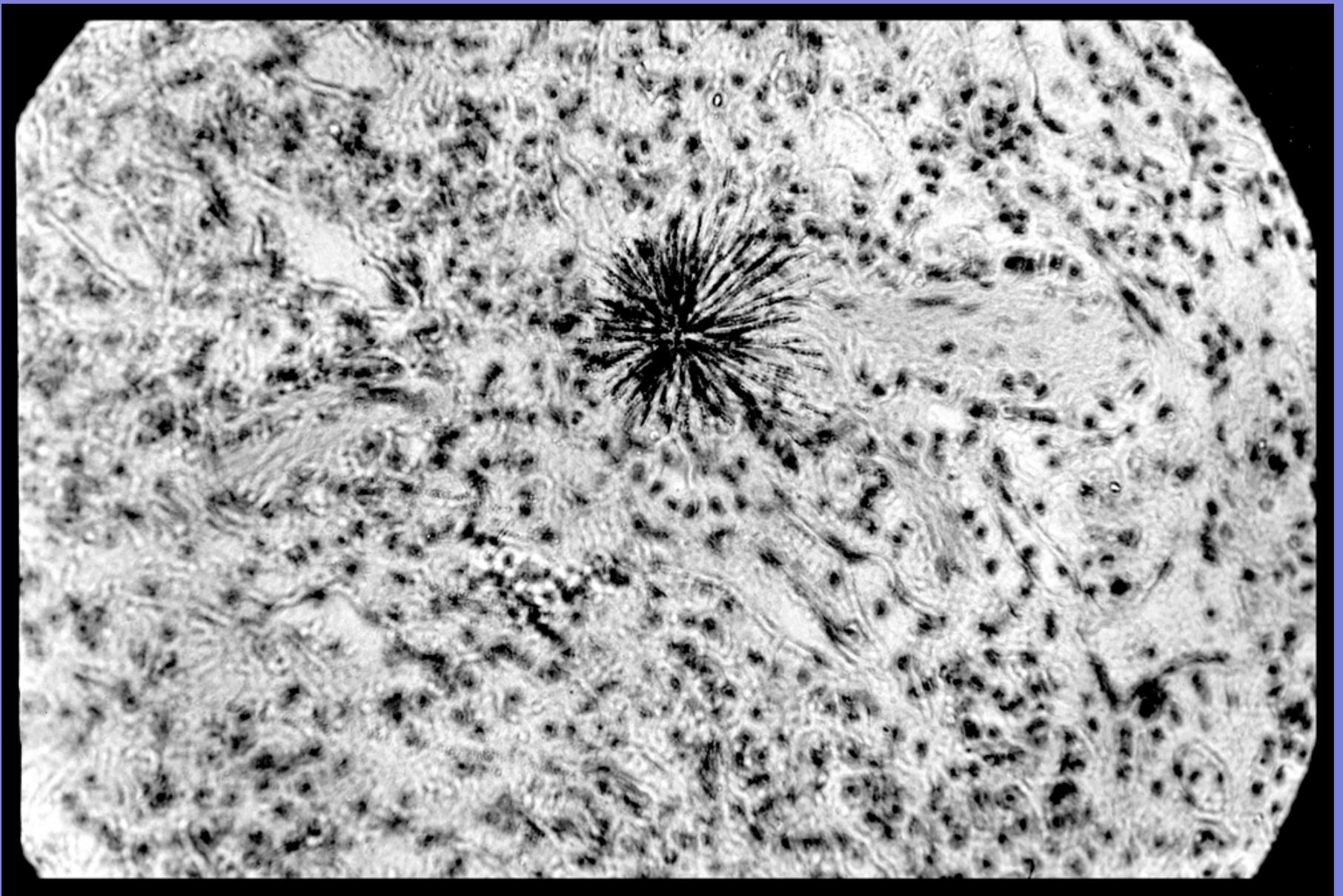
1998

Direction régionale de la santé publique



Underground Miner (Navajo)

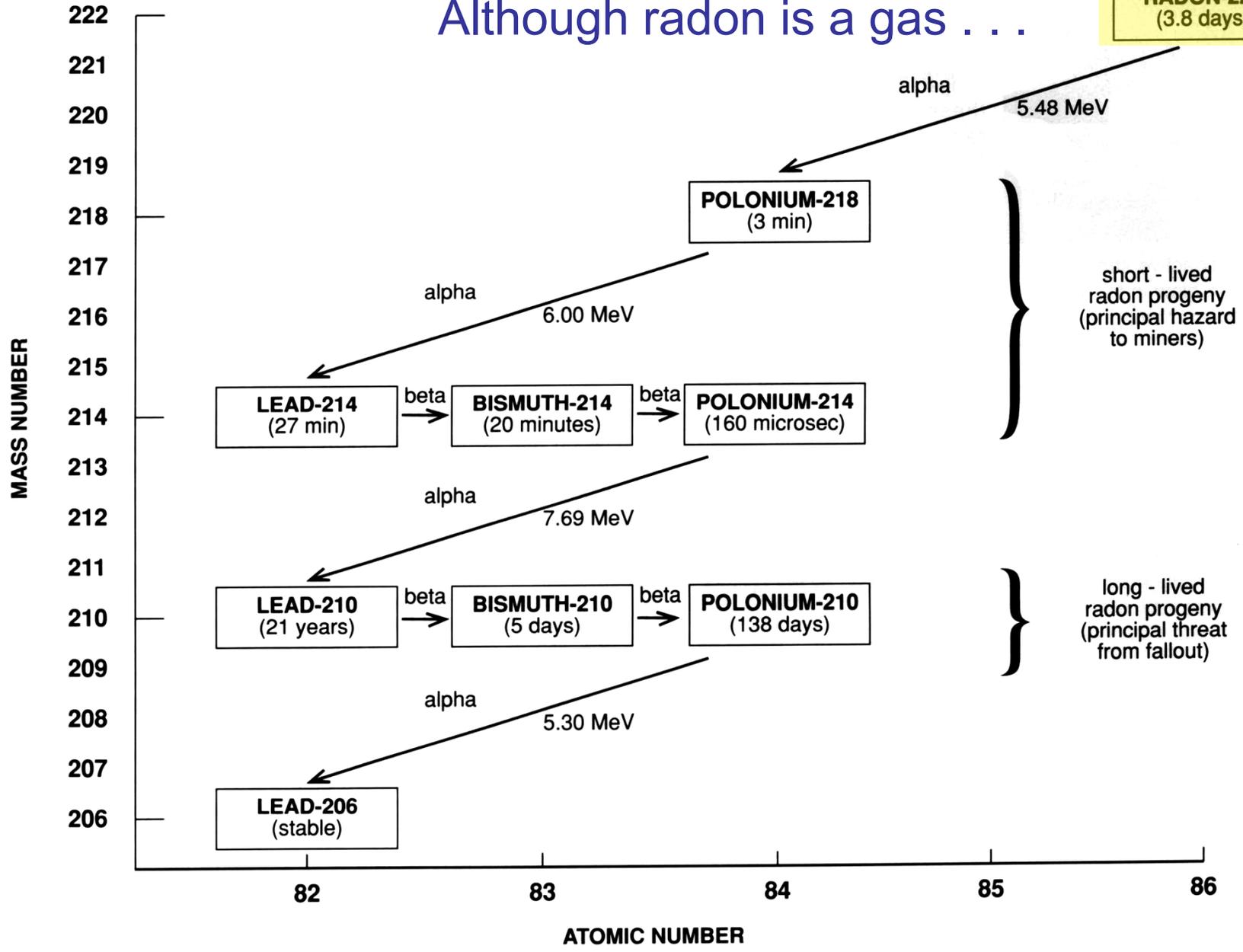
Radon



“Alpha Radiation” from a tiny radioactive particle in lung tissue

Although radon is a gas . . .

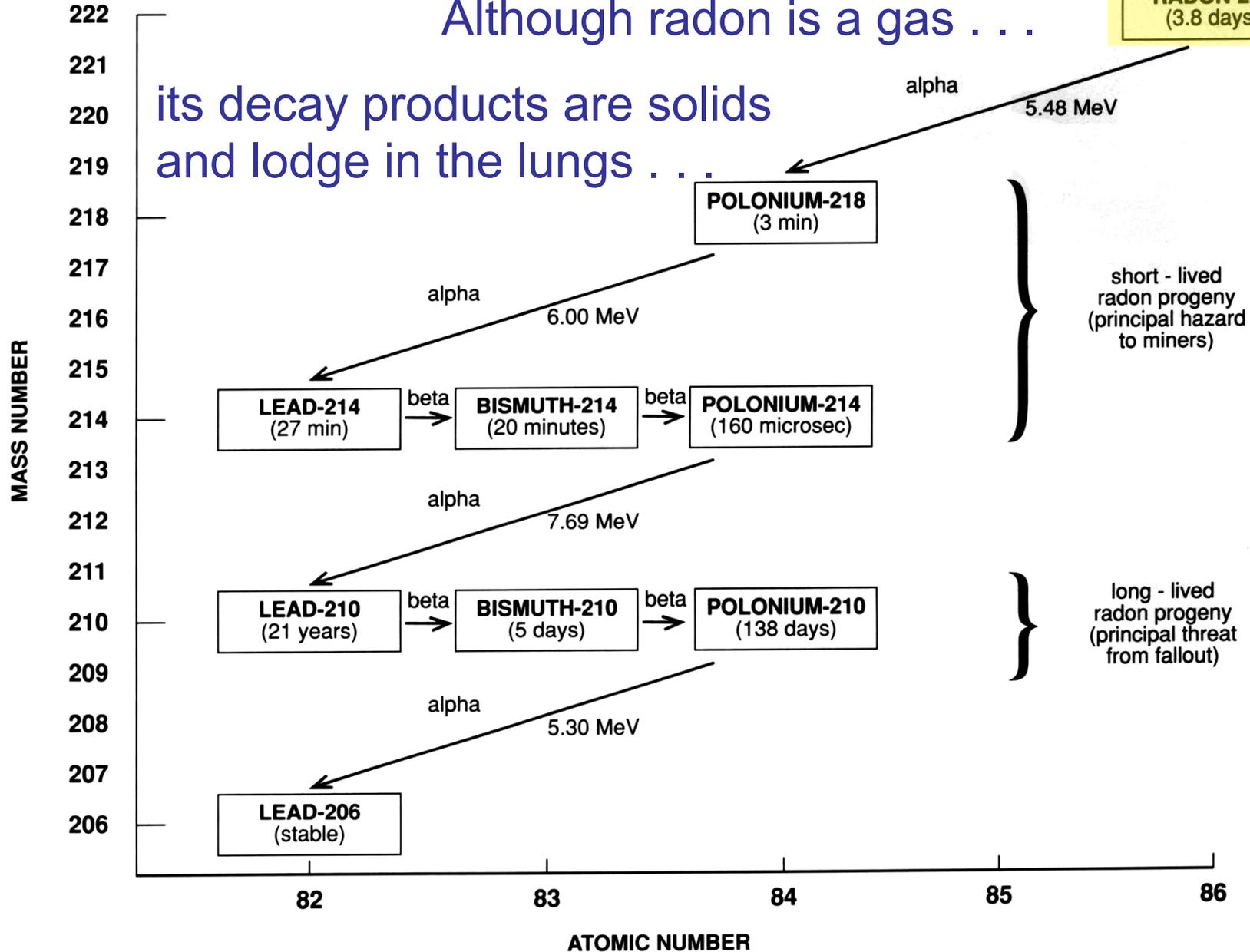
RADON-222
(3.8 days)



Although radon is a gas . . .

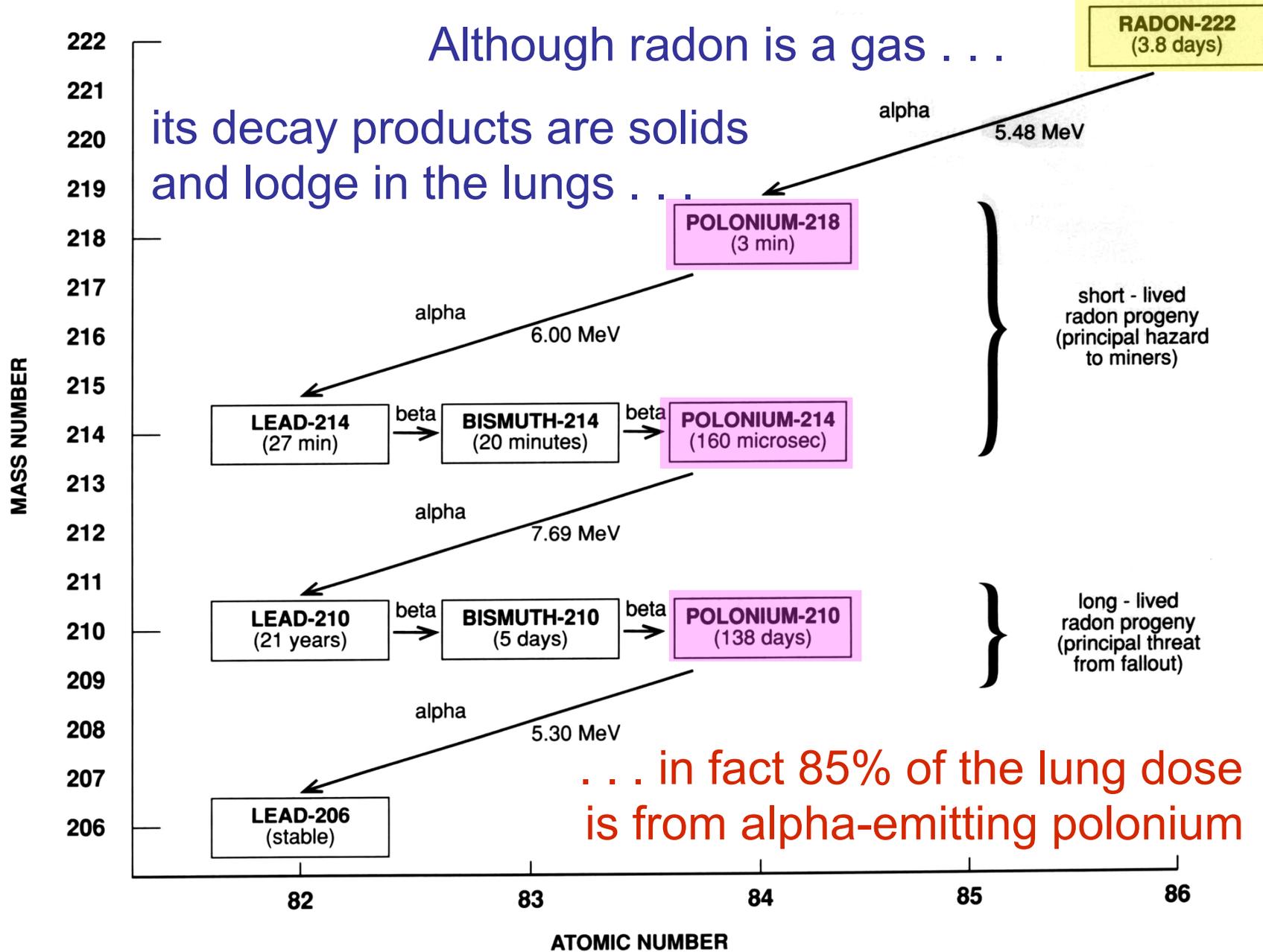
its decay products are solids
and lodge in the lungs . . .

RADON-222
(3.8 days)



Although radon is a gas . . .

its decay products are solids
and lodge in the lungs . . .



. . . in fact 85% of the lung dose
is from alpha-emitting polonium

WASHINGTON (June 2007) — A preponderance of scientific evidence shows that **even low doses of ionizing radiation, such as gamma rays and X-rays, are likely to pose some risk of adverse health effects**, says a new report from the National Academies' National Research Council

2007 press release announcing publication of the National Academy of Sciences' BEIR-VII Report on the Biological Effects of Ionizing Radiation

. . . such radiation **can cause DNA damage that eventually leads to cancers**. However, more research is needed to determine whether **low doses** of radiation may also cause other health problems, such as heart disease and stroke, which are now seen with high doses

*2007 press release announcing
publication of the National Academy of
Sciences' BEIR-VII Report on the
Biological Effects of Ionizing Radiation*

"The scientific research base shows that there is no threshold of exposure below which low levels of ionizing radiation can be demonstrated to be harmless or beneficial," said committee chair Richard R. Monson, Harvard School of Public Health, Boston.

2007 press release announcing publication of the National Academy of Sciences' BEIR-VII Report on the Biological Effects of Ionizing Radiation

CANADA
DEPARTMENT OF MINES

INVESTIGATIONS IN ORE DRESSING AND METALLURGY

1931
OTTAWA

PRECAUTIONS FOR WORKERS IN THE
TREATMENT OF RADIUM ORES

W. R. McClelland

The hazards involved in the handling of high grade radioactive materials make necessary the adoption of certain precautions. Recent investigations in the field of radium poisoning have led to the conclusion that precautions are necessary even in the handling of substances of low radioactivity. The ingestion of small amounts of radioactive dust or emanation over a long period of time will cause a build up of radioactive material in the body, which eventually may have serious consequences. lung cancer, bone necrosis and rapid anemia are possible diseases due to deposition of radioactive substances in the cell tissue or bone structure of the body.

Fact #3

Radioactivity
also endangers
the environment

International Physicians for the Prevention of Nuclear War

(IPPNW) August 2010 [Nobel Prize 1985]

- Uranium ore mining . . . represents a grave threat to health and to the environment . . . and an incalculable risk for world peace and disarmament.
- IPPNW issues a global call to action for a ban on uranium mining worldwide.

Basel, Switzerland, August 29 2010

Warnings From the EPA

“. . . mining and refining processes can **introduce radionuclides . . . into the environment** at unnaturally high rates.”

“This makes the handling of materials associated with Rare Earth element production of utmost concern. **Extreme care should be used to ensure . . . contaminants are not released** into the environment.”

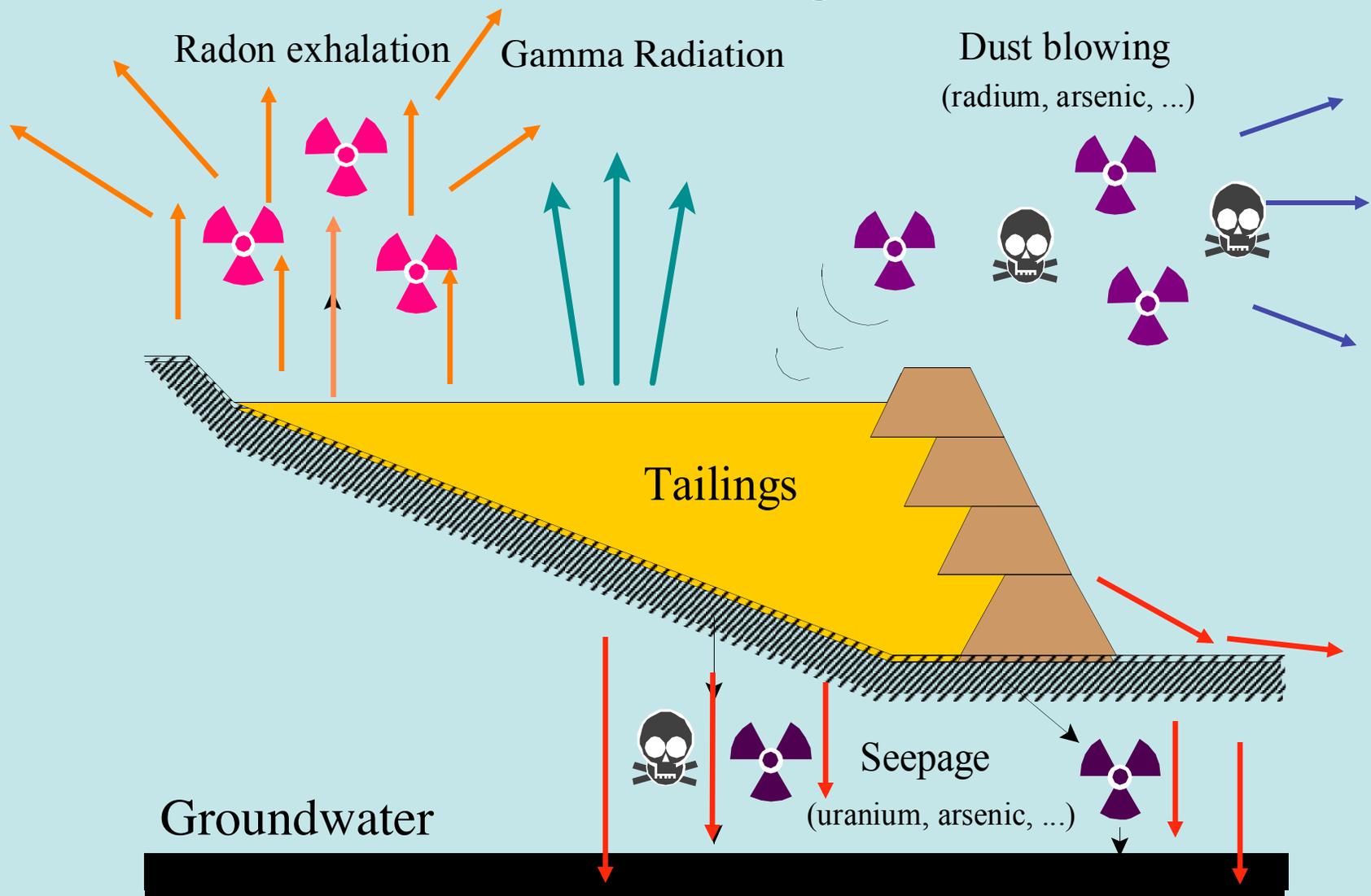
US EPA, Investigating Rare Earth Element Mine Development, 2011



*Photo:
Robert Del Tredici*

Behind this 10 meter wall is 70 million tonnes of Uranium tailings.

Mill Tailings Hazard



Warnings From the EPA

“Ores containing uranium-238 and thorium-232 are very mobile as dust resulting in air and soil contamination, [from] where radon-222 gas is constantly released”

“Stories of environmental pollution and human sickness remain frequent in areas near Chinese Rare Earth element production facilities.”

US EPA, Investigating Rare Earth Element Mine Development, 2011

Fact #4

There are many
unanswered
questions

Warnings From the EPA

“Unless the ores and wastes are carefully collected and monitored, **contaminants could be easily introduced into the environment.**”

“In fact, it is more likely the possible contaminants from **waste rock piles** will be introduced into the environment considering **these possible contaminants are not contained in a pit . . .**”

US EPA, Investigating Rare Earth Element Mine Development, 2011

Key Questions

- 1. How will radioactive wastes be safely secured?*
- 2. Who will pay for monitoring and remediation?*
- 3. Will advanced milling techniques be required?*
- 4. Will human use of radwastes be prevented?*
- 5. Will wildlife and food chain be checked?*
- 6. Will Quebec get needed radiological expertise?*
- 7. Will children be educated on radioactive risks?*

The End

of a backgrounder prepared by
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